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Stoichiometric Calculations are Based on Chemical Formulas  
Let's learn some terms used in Stoichiometry first. Formula Mass: It is the sum of the atomic weights of the various atoms present in the molecule of the substance. For example, we can calculate the formula mass of Na<sub>2</sub>S as 2 (23) + 1 (32) = 78

## **Stoichiometry and Stoichiometric Calculations: Concepts**

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Stoichiometry, by definition, is the calculation of the quantities of reactants or products in a chemical reaction using the relationships found in the balanced chemical equation. The word stoichiometry is actually Greek from two words:  $\sigma\tau\omicron\iota$   $\iota\omicron\mu\omicron\nu$  ( $\sigma\tau\omicron\iota$   $\iota\omicron\mu\omicron\nu$ ), which means "element", and  $\mu\epsilon\tau\epsilon\omicron\varsigma$  ...

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